

Effective Date of Contract: April 1, 1974

Contract Expiration Date: June 30, 1976

Amount of Contract: \$159,892
Form Approved, Budget -- No. 22-R0293

The views and conclusions contained in this document are those of the authors and should not be interpreted as necessarily representing the official policies, either expressed or implied, of the Advanced Research Projects Agency of the U. S. Government.

DISTRIBUTION STATEMENT A
Approved for public releases
Distribution Unlimited

374 625

ACCESSION TO:

HTIS TABLE CALL OF THE COLOR OF THE CATTON CATTON OF THE CATTON OF THE

EFFECT OF THE MgO/SiO2 RAT10 ON THE STRENGTH OF HOT-PRESSED Si3N4

Technical Report #3, October 15, 1975

Westinghouse Electric Corporation Research and Development Center

Contract Number N00014-74-C-0284

Sponsored by the Advanced Projects Agency ARPA Order Number 2697 Program Code Number 01269

Scientific Officer: Dr. A. M. Diness Office of Naval Research

Principal Investigator: Dr. F. F. Lange (412) 256-3684

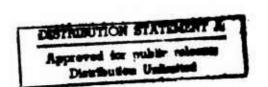
Effective Date of Contract: April 1, 1974

Contract Expiration Date: June 30, 1976

Amount of Contract: \$159,892

Form Approved, Budget -- No. 22-R0293

The views and conclusions contained in this document are those of the authors and should not be interpreted as necessarily representing the official policies, either expressed or implied, of the Advanced Research Projects Agency of the U. S. Government.



October 15, 1975 EFFECT OF THE MgO/SiO2 RATIO ON THE STRENGTH OF HOT-PRESSED Si3N4 C. A. Andersson, F. F. Lange and J. L. Iskoe Metallurgy and Metals Processing Westinghouse Research Laboratories Pittsburgh, Pennsylvania ABSTRACT The strength of hot-pressed  $Si_3N_4$  containing different  ${\rm Mg0/Si0_2}$  molar ratios was studied. Significant increases in strength at  $1400^{\circ}$ C were observed as the Mg0/Si02 ratio increased to 3-4. Smaller decreases in strength at both 25°C and 1400°C were observed when  $Mg0/Si0_2 > 4$ . These observations are briefly discussed with regard to possible chemistry changes of the grain boundary phase believed to be present in hot-pressed Si<sub>3</sub>N<sub>4</sub>. no.4 p. 1 INTRODUCTION Silicon nitride hot-pressed with the aid of MgO is believed to consist of  $\beta$ -Si<sub>3</sub>N<sub>4</sub> grains separated by a silicate phase. [1] Impurities such as CaO have been shown to effect the mechanical properties at elevated temperatures by reducing the refractoriness of the grain boundary phase. [2] Recognizing that a significant improvement in these properties could be achieved by the reduction of such contaminants, a successful effort was made to produce higher purity starting powder at Westinghouse. Once the overwhelming effect of impurities was radically reduced, a new effect was discovered. Chronologically, it was discovered that the high temperature strength was dependent on the amount of the

MgO hot-pressing aid used to densify a given starting powder as shown in Fig. 1. This unexpected effect was eventually rationalized by hypothesizing that the MgO/SiO $_2$  ratio\* was the controlling factor in this apparent relation. As detailed below, this hypothesis was confirmed by testing materials fabricated with different MgO/SiO $_2$  ratios.

## 2. EXPERIMENTAL

## 2.1 Powders

Different Si $_3N_4$  powders were manufactured by nitriding Si powder with additions of 0.0, 1.0 and 3.0 wt/o Si0 $_2$ . The phase content of the resulting powders were 83-93%  $\alpha$ -Si $_3N_4$ , 17-7%  $\beta$ -Si $_3N_4$  and  $\leq$  1% Si as determined by X-ray diffraction analysis. No correlation could be made between the Si0 $_2$  additions and these crystalline phases.

The oxygen content of representative powders was determined after nitriding with the inert gas fusion, thermoconductivity method. Several powders were also analyzed by neutron activation, courtesy of H. Priest, AMMRC. Figure 2 illustrates that the oxygen content of different batches of  $\mathrm{Si}_3\mathrm{N}_4$  powder is a function of the  $\mathrm{Si0}_2$  added prior to nitriding. The oxygen content of the Si powder was 0.4-0.5 wt/o. Since the oxygen content of the Si powder is most likely due to a surface layer of  $\mathrm{Si0}_2$ , and the increase in oxygen content was proportional to the  $\mathrm{Si0}_2$  added prior to nitriding it was assumed that the oxygen content after nitriding corresponded to the  $\mathrm{Si0}_2$  in the  $\mathrm{Si}_3\mathrm{N}_4$  powder.\*\* It should be noted that the molar content of the apparent  $\mathrm{Si0}_2$  ranged between 1.7 and 6.7 mole/o. Commercial  $\mathrm{Si}_3\mathrm{N}_4$  powders can have significantly higher  $\mathrm{Si0}_2$  contents. [4]

<sup>\*</sup> SiO2 is a common contaminant in Si<sub>3</sub>N<sub>4</sub> powders.[3,4]

<sup>\*\*</sup> As recently shown by Kohatsu and McCauley [5] oxygen is not a necessary element of the  $\alpha$ -Si3N4 structure as previously reported by Wild, et al. [6]

Table 1 reports the cation impurity content of the  ${\rm Si}_3{}^{\rm N}_4$  powders. No variation in impurity content outside of experimental error was observed between different batches.

 ${
m Si}_3{
m N}_4$  powders containing different MgO/SiO $_2$  ratios were prepared by mixing 2-8 w/o MgO with  ${
m Si}_3{
m N}_4$  powders containing different apparent  ${
m SiO}_2$  contents. In addition, the MgO/SiO $_2$  ratio of several powders was also varied by mixing both MgO and  ${
m SiO}_2$  into a powder with an oxygen content of 0.4 w/o (0.75 w/o  ${
m SiO}_2$ ). Mixing and particle size reduction was performed by milling with methanol in polyethylene bottles using WC cylindrical grinding media. Oxygen analysis before and after milling showed no changes that could not be accounted for by the MgO addition.

# 2.2 Densification and Mechanical Testing

After stir-drying, the milled composite powders were hot-pressed in a  $\rm N_2$  atmosphere with a stress of 28 MN/m $^2$  at a temperature of 1750°C for 1-4 hrs to produce 5 cm diameter by 0.75 cm discs. The powders were separated from the graphite die with grafoil. Density was determined by water immersion.

Bar specimens (0.317 x 0.635 x 3.17 cm) were sectioned and ground with a 320 grit diamond wheel. Room temperature flexural strength measurements were made at a crosshead speed of 0.05 cm/min using a metal fixture (0.635 cm inner and 1.905 cm outer loading spans). Elevated temperature measurements were performed in air at 1400°C with a crosshead speed of 0.005 cm/min using a hot-pressed  $\rm Si_3N_4$  fixture (0.950 cm inner and 2.22 cm outer loading spans).

#### RESULTS

Densities varied between 3.20 gm/cc and 3.29 gm/cc due to a slight variation in the tungsten contamination during milling.

Flexural strength data at  $25^{\circ}$ C and  $1400^{\circ}$ C are shown in Fig. 3 as a function of the MgO/SiO<sub>2</sub> ratio. Each point, represented by a number indicating the weight % of the MgO hot-pressing aid, corresponds to the average result from a single billet (3-6 measurements at  $25^{\circ}$ C and 3 measurements at  $1400^{\circ}$ C). Because the oxygen content of only half of the  $Si_3N_4$  powders were directly determined, the linear relation shown in Fig. 2 was used to calculate the oxygen content of the other powders. The oxygen content was used to calculate the apparent  $SiO_2$  content and thus, the MgO/SiO<sub>2</sub> molar ratio. The circled numbers correspond to materials where  $SiO_2$  was added to  $Si_3N_4$  powder.

As shown, the sarength at 25°C exhibits much scatter.\* Mean values are 655 MN/m<sup>2</sup> for Mg0/Si0<sub>2</sub>  $\leq$  4, apparently decreasing to 520 MN/m<sup>2</sup> at Mg0/Si0<sub>2</sub> = 10.

At  $1400^{\circ}\text{C}$ , the mean strength increases from  $170~\text{MN/m}^2$  at  $10\text{w}~\text{Mg0/Si0}_2$  ratios to  $415~\text{MN/m}^2$  at  $10\text{Mg0/Si0}_2$  ratios to  $10\text{Mg0/Si0}_2$  ratios. The circled points should be noted since they represent results for materials where the  $10\text{Mg0/Si0}_2$  ratio was 10Mg adding 10Mg to the starting 10Mg powder. Without the additional 10Mg, the same 10Mg powders had greater 10Mg powders and corresponding higher strengths.

Figure 4 illustrates load-deflection curves for selected specimens with different  ${\rm Mg0/Si0_2}$  ratios, indicating that less non-elastic deformation occurs at  ${\rm Mg0/Si0_2}$  > 3.

#### 4. DISCUSSION

The increase in strength at  $1400^{\circ}\text{C}$  for  $0 < \text{Mg0/Si0}_{2} \le 4$  can be interpreted as due to a change in the grain boundary phase. Either a change in its refractoriaess or its volume content would result in

<sup>\*</sup> The observed scatter may have been due to the small differences in hot-pressing schedules, e.g., heating rates, mode of pressure application, etc., used by the 3 different investigators, which can influence the resulting microstructure and thus the fracture toughness and strength of hot-pressed Si<sub>3</sub>N<sub>4</sub>.[7]

lower rates for grain boundary separation and sliding, [8] thus decreasing the materials susceptibility to sub-critical crack growth and non-elastic deformation at high temperatures. [9]

The reason for the apparent decrease in strength at  $25^{\circ}$ C and  $1400^{\circ}$ C for  $MgO/SiO_2 > 4$  requires further experiments. The authors believe that this effect is related to the volume content of the grain boundary phase, which might govern the material's grain structure.

The composition and volume fraction of the grain boundary phase are presently unknown. Oyama and Kamigaito  $^{[10]}$  have reported data suggesting that up to 30 mole % of MgO is soluble in  $\mathrm{Si}_3\mathrm{N}_4$ . These investigators neglected the presence of the  $\mathrm{SiO}_2$  contamination in the  $\mathrm{Si}_3\mathrm{N}_4$  powder; thus the real phase relations in the  $\mathrm{Si}_3\mathrm{N}_4$ - $\mathrm{SiO}_2$ -MgO-Mg $_3\mathrm{N}_2$  system are presently unknown. In any case, Oyama and Kamigaito's results suggest that the MgO densification aid would be present in both the  $\beta$ -Si $_3\mathrm{N}_4$  structure and the grain boundary phase.

A change in either the refractoriness or the volume content of the grain boundary can occur by shifting compositions with the  $\mathrm{Si}_3\mathrm{N}_4\mathrm{-Si0}_2\mathrm{-Mg}_0\mathrm{-Mg}_3\mathrm{N}_2$  system, e.g., by shifting to a new phase field, the grain boundary phase may not be a silicate. On the other hand, the change in strength with the Mg0/Si02 ratio may be a result of shifting away from glass forming regions in the Mg0-Si02-impurity system, i.e., silicate composition containing large proportions of Mg0 are less apt to form glasses upon solidification. Phase equilibrium studies would clarify these different viewpoints.

#### 5. CONCLUSION

It has been shown that the  ${\rm Mg0/Si0}_2$  ratio has a significant effect on the high temperature strength of  ${\rm Si}_3{\rm N}_4$ . Thus optimum strengths can only be achieved by controlling extrinsic impurities, e.g., CaO, and the  ${\rm SiO}_2$  content of starting  $\alpha{\rm -Si}_3{\rm N}_4$  powders.

#### **ACKNOWLEDGMENTS**

The authors would like to acknowledge the technical assistance of J. J. Nalevanko, W. J. Carmen and J. P. Yex. This work was supported by the Advanced Projects Research Agency through the Office of Naval Research, Contract No. N00014-74-C-0284.

#### REFERENCES

- 1. F. F. Lange, "Strong, High-Temperature Ceramics", Ann. Rev. Mat. Sci. 4, 365 (1974).
- 2. J. L. Iskoe, F. F. Lange and E. S. Diaz, "Effect of Selected Impurities on the High Temperature Mechanical Properties of Hot-Pressed Si<sub>3</sub>N<sub>4</sub>", J. Mat. Sci. (in press).
- 3. I. Colquhoun, D. P. Thompson, W. I. Wilson, P. Grieveson and K. Jack, "Determination of Surface  $SiO_2$  and Its Effect on the Hot-Pressing Behavior of  $\alpha$ -Si $_3N_4$  Powder", <u>Proc. Brit. Ceram. Soc. No. 22</u>, 181 (1973).
- 4. F. F. Lange, "Task I: Fabrication, Microstructure and Selected Properties of SiAlON Compositions", Final Rept. NAVAIR Contract N00019-73-C-0208, February 1974.
- 5. I. Kohatsu and J. W. McCauley, "Re-Examination of the Crystal Structure of  $\alpha$ -Si<sub>3</sub>N<sub>A</sub>", Mat. Res. Bull. 9, 917 (1974).
- 6. S. Wild, P. Grieveson and K. H. Jack, "Crystal Structures of  $\alpha$  and  $\beta$  Si  $_3N_4$  and Ge  $_3N_4$ ", Special Ceramics, 5, Ed. by P. Popper, p. 385, B.C.R.A., Stoke-on-Trent (1974).
- 7. J. L. Iskoe and F. F. Lange, "Microstructural Development of  ${\rm Si}_3{\rm N}_4$  Hot-Pressed with MgO", to be published.
- 8. F. F. Lange, "Non-Elastic Deformation of Polycrystals with a Liquid Boundary Phase", <u>Deformation of Ceramic Materials</u>, Ed. by R. C. Bradt and R. E. Tressler, p. 361, Plenum (1975).

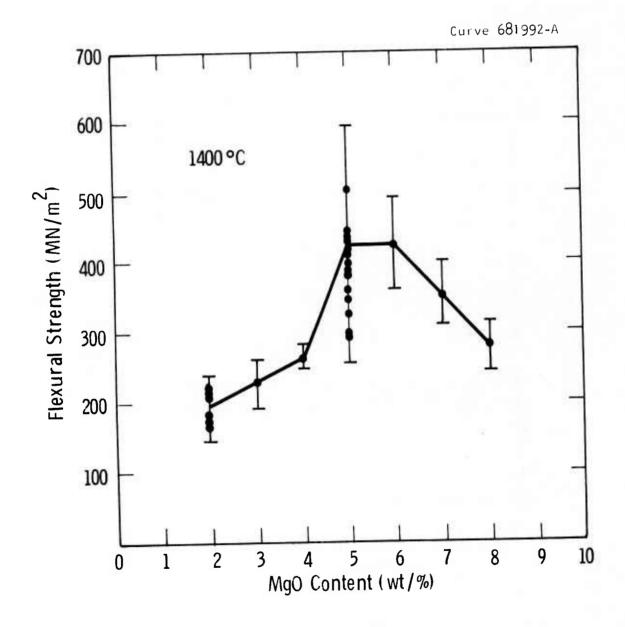
- 9. F. F. Lange, "High-Temperature Strength Behavior of Hot-Pressed Si<sub>3</sub>N<sub>4</sub>: Evidence for Sub-Critical Crack Growth", <u>J. Am. Ceram. Soc.</u> 57, 84 (1974).
- 10. Y. O. Oyama and O. Kamigaito, "A Study on the Sintered Si<sub>3</sub>N<sub>4</sub>-MgO System", Yogyo-Kyokai-Shi <u>81</u>, 34 (1973).

## FIGURE CAPTIONS

- F1G. 1 -- Flexural strength at 1400°C vs Mg0 content used for densification. Each point is an average of 3 specimens from a single billet. Test conditions are given in text. Oxygen content of different powders was 0.9-1.6 w/o.
- F1G. 2 -- Oxygen content of Si<sub>3</sub>N<sub>4</sub> powder vs SiO, added to Si powder prior to nitriding.
- FIG. 3 -- Flexural strength vs MgO/SiO2 molar ratio.
- FIG. 4 -- Bending moment vs deflection for typical flexural specimens fabricated with different Mg $0/\mathrm{Si}\,0_2$  molar ratios.

 $\label{thm:continuous} Tab \ le \ 1$  Spectrographic Analyses of Westinghouse Si  $_3N_4$  Starting Powder (wt %)

A1		0.08
Ag	<	0.001
В		0.001
Ca		0.016
Cr		0.01
Fe	>	0.1
Mg		0.001
Mn		0.05
Мо	<	0.003
Ni	<	0.01
Pb	<	0.01
-		
Pb	<	0.01
Pb Sb	< <	0.01
Pb Sb Sn	< <	0.01 0.01 0.01



A STREET, P

Andreas I

Mary Control of the C

Fig. 1

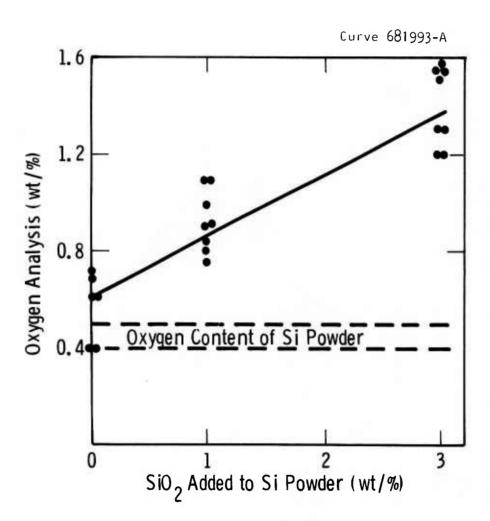


Fig. 2

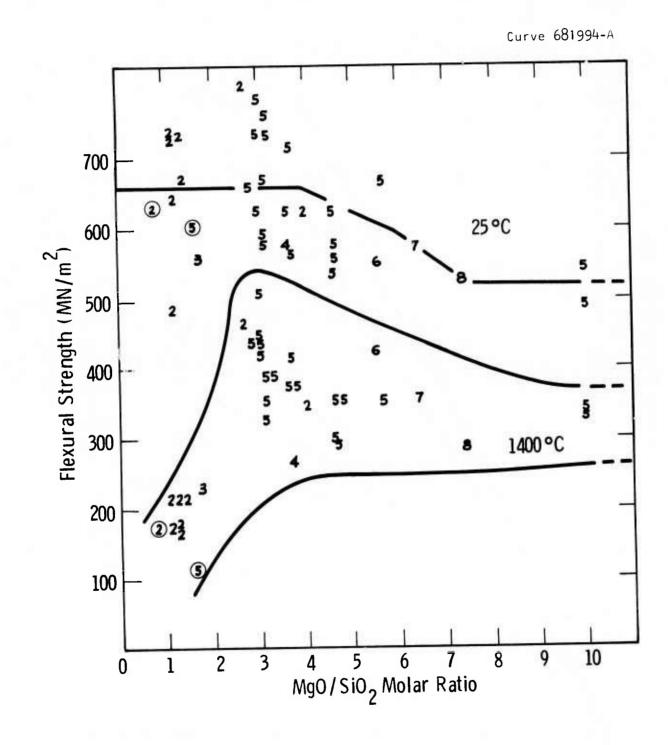


Fig. 3

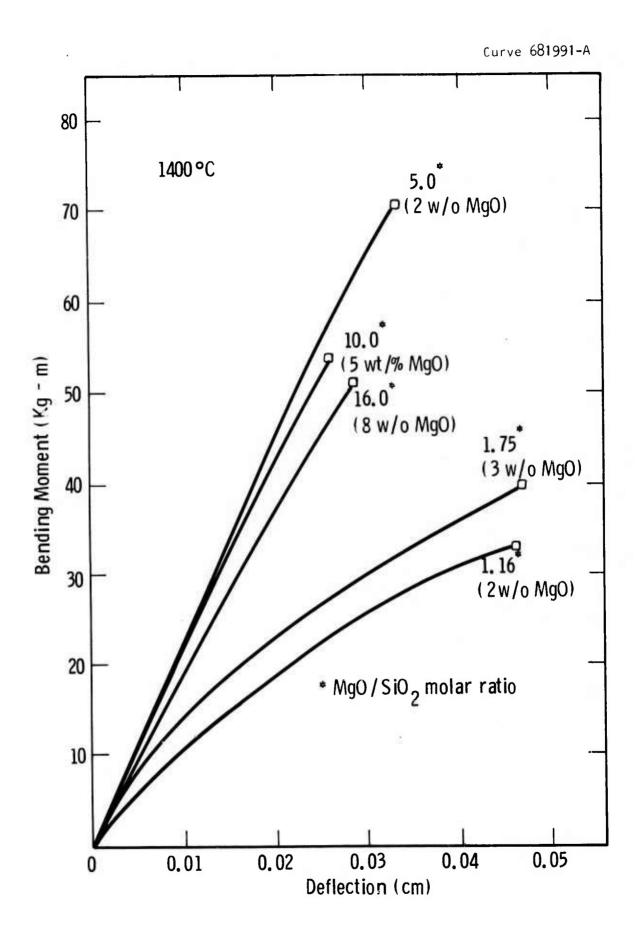


Fig. 4

Security Classification			
	TROL DATA - R&		111
(Security classification of title, body of abstract and indexing	g annotation must be en		he overall report is classified)
1. ORIGINATING ACTIVITY (Corporate author)			assi fied
Westinghouse Research & Development Ce	enter	2 b GROUE	
Pittsburgh, PA 15235		25 GROOM	
3 REPORT TITLE		<u> </u>	
	PREMOTH OF HOT	_DDFSSET	Si aNa
EFFECT OF THE MgO/SiO2 RATIO ON THE ST	IKENGIII OF HOT	-I KLOULL	, 513.14
4 DESCRIPTIVE NOTES (Type of report and inclusive dates)			
(C) (I the second district)			
5 AUTHOR(S) (Last name, first name, initial)			
Andersson, C. A., Lange, F. F. and Isl	koe, J. L.		l i
6. REPORT DATE	7a. TOTAL NO. OF F	AGES	7b. NO. OF REFS
October 15, 1975			
B.B. CONTRACT OR GRANT NO.	9 e. ORIGINATOR'S R	EPORT NUM	BER(S)
N00014-74-C-0284			
b. PROJECT NO.			
	95 OTHER REPORT	NO(S) (Any	other numbers that may be easigned
с.			W SEE
d.	None		
10 AVAILABILITY/LIMITATION NOTICES			
Reproduction in whole or in part is p	ermitted for a	ny purp	ose of the
U. S. Government, Distribution of thi	s document is	UNLIMIT	ED.
A SUPER TURNITARY NATES	12. SPONSORING MIL	ITARY ACT	IVITY
11. SUPPLEMENTARY NOTES			
	Advanced Res	earch P	rojects Agency
13. ABSTRACT			
The strength of hot-pressed Si <sub>3</sub> N <sub>4</sub> con	taining differ	ent Mg0	/Si0 <sub>2</sub> molar ratios
was studied. Significant increases i	n strength at	1400°C	were observed as
the MgO/SiO <sub>2</sub> ratio increased 3-4. Sm			
25°€ and 1400°C were observed when Mg	$0/\sin 2 > 4.$	These ob	servations are
briefly discussed with regard to poss	ible chemistry	change	s of the grain
boundary phase believed to be present	. In not-press	ou 013114	•
ļ			
1			
1			
1			

DD 1 FORM 1473

Unclassified
Security Classification

Security Classification	LIN	LINK A		LINK B		LINKC	
KEY WORDS		ROLE	wT	ROLE	wT	ROLE	WT
silicon n	itride						
hot							
fabricati	on		1				
strength		ı.					
creep							
boundarie	es						
phase							
pressing							
grains							
equilibri	1100		J		1		1

- I. ORIGINATING ACTIVITY: Enter the name and address of the contractor, subcontractor, grantee, Department of Defense activity or other organization (corporate author) issuing the report.
- 2a. REPORT SECURITY CLASSIFICATION: Enter the overall security classification of the report. Indicate whether "Restricted Data" is included. Marking is to be in accordance with appropriate security regulations.
- 2b. GROUP: Automatic downgrading is specified in DoD Directive 5200. IO and Armed Forces Industrial Manuel. Enter the group number. Also, when applicable, show that optional markings have been used for Group 3 and Group 4 as authorized.
- 3. REPORT TITLE: Enter the complete report title in all capital letters. Titles in all cases should be unclassified. If a meaningful title cannot be selected without classification, show title classification in all capitals in parenthesis immediately following the title.
- 4. DESCRIPTIVE NOTES: If appropriate, enter the type of report, e.g., interim, progress, summary, annual, or final. Give the inclusive dates when a specific reporting period is covered.
- 5. AUTHOR(S): Enter the name(s) of author(s) as shown on or in the report. Enter last name, first name, middle initial. If military, show rank and branch of service. The name of the principal author is an absolute minimum requirement.
- 6. REPORT DATE: Enter the date of the report as day, month, year, or month, year. If more than one date appears on the report, use date of publication.
- 7a. TOTAL NUMBER OF PAGES: The total page count should follow normal pagination procedures, i.e., enter the number of pages containing information.
- 7b. NUMBER OF REFERENCES: Enter the total number of references cited in the report.
- 8a. CONTRACT OR GRANT NUMBER: If appropriate, enter the applicable number of the contract or grant under which the report was written.
- 8b, 8c, & 8d. PROJECT NUMBER: Enter the appropriate military department identification, such as project number, subproject number, system numbers, task number, etc.
- 9a. ORIGINATOR'S REPORT NUMBER(S): Enter the official report number by which the document will be identified and controlled by the o.iginating activity. This number must be unique to this report.
- 9b. OTHER REPORT NUMBER(S): If the report has been assigned any other report numbers (either by the originator or by the sponsor), also enter this number(s).
- IO. AVAILABILITY/LIMITATION NOTICES: Enter any limitations on further dissemination of the report, other than those

imposed by security classification, using standard statements such as:

- (I) "Qualified requesters may obtain copies of this report from DDC."
- (2) "Foreign announcement and dissemination of this report by DDC is not authorized."
- (3) "U. S. Government agencies may obtain copies of this report directly from DDC. Other qualified DDC users shall request through
- (4) "U. S. military agencies may obtain copies of this report directly from DDC. Other qualified users shall request through
- (5) "AII distribution of this report is controlled. Qualified DDC users shall request through

If the report has been furnished to the Office of Technical Services, Department of Commerce, for sale to the public, indicate this fact and enter the price, if known

- II. SUPPLEMENTARY NOTES: Use for additional explana-
- 12. SPONSORING MILITARY ACTIVITY: Enter the name of the departmental project office or laboratory sponsoring (paying for) the research and development. Include address.
- 13. ABSTRACT: Enter an abstract giving a brief and factual summary of the document indicative of the report, even though it may also appear elsewhere in the body of the technical report. If additional space is required, a continuation sheet shall be attached.

It is highly desirable that the abstract of classified reports be unclassified. Each paragraph of the abstract shall end with an indication of the military security classification of the information in the paragraph, represented as (TS). (S). (C), or (U).

There is no limitation on the length of the abstract. However, the suggested length is from 150 to 225 words.

14. KEY WORDS: Key words are technically meaningful terms or short phrases that characterize a report and may be used as index entries for cataloging the report. Key words must be selected so that no security classification is required. Identifiers, such as equipment model designation, trade name, military project code name, geographic location, may be used as key words but will be followed by an indication of technical context. The assignment of links, rules, and weights is optional.

# INTERRELATION BETWEEN CREEP AND SLOW CRACK GROWTH

Technical Report #4, October 15, 1975

Westinghouse Electric Corporation Research and Development Center

Contract Number N00014-74-C-0284

Sponsored by the Advanced Projects Agency ARPA Order Number 2697 Program Code Number 01269

Scientific Officer: Dr. A. M. Diness

Office of Naval Research

Principal Investigator: Dr. F. F. Lange

(412) 256-3684

Effective Date of Contract: April 1, 1974

Contract Expiration Date: June 30, 1976

Amount of Contract: \$159,892

Form Approved, Budget -- No. 22-R0293

The views and conclusions contained in this document are those of the authors and should not be interpreted as necessarily representing the official policies, either expressed or implied, of the Advanced Research Projects Agency of the U. S. Government.

October 15, 1975

INTERRELATIONS BETWEEN CREEP AND SLOW CRACK GROWTH

F. F. Lange

Metallurgy and Metals Processing
Westinghouse Research Laboratories
Pittsburgh, Pennsylvania 15235

ABSTRACT

No.4. Phenomenological interrelations between creep strain and slow crack growth are derived based on the fact that the total creep strain at fracture is determined by the growth kinetics of a pre-existing crack. It is shown that measurements of the creep strain in a region remote from the growing crack can be used to predict failure periods for materials that concurrently exhibit creep and slow crack growth.

ho. 5 10 1

# 1. INTRODUCTION

Certain materials have been found to concurrently exhibit creep and slow crack growth. Examples are discalloy, AISI-4340 tempered steel, [2] glass-ceramics [3] and hot-pressed Si<sub>3</sub>N<sub>4</sub>. [4] For a given material and condition, both phenomenon can be governed by the same mechanism, e.g., grain boundary separation and sliding is the mechanism governing both creep and slow crack growth in polycrystals with a viscous grain boundary phase. [5] Or, both phenomenon can be governed by different mechanisms, e.g., creep, by dislocation movement and slow crack growth, by an environmental, stress-corrosion mechanism.

Usually creep and slow crack growth are treated independently. That is, creep laws are obtained for a material to define non-elastic strain as a function of stress and time and slow crack growth behavior is obtained from crack velocity-stress intensity relations that are used to predict failure time as a function of stress history. The purpose of this article is to show that these independently measured laws are interrelated regardless of the mechanism that control either phenomenon. It will be shown that the total strain at failure depends on the stress laws governing both phenomenon and that a measurement of strain and/or strain rate can be used to predict the failure time due to slow crack growth.

### 2. ANALYSIS OF INTERRELATION

It is assumed that a body under stress  $(\sigma_a)$  simultaneously exhibits non-elastic strain (creep) and slow crack growth. Also, for simplicity, it is assumed\* that the creep strain can be described by

$$\varepsilon = \varepsilon_0 + A \sigma_a^n t,$$
 (1)

and that slow crack growth can be described by the crack velocity (v) - stress intensity factor (K) relation

$$v = B K^{m}, (2)$$

where  $\epsilon_0$ , A, n, B, and m are constants at a given temperature. The constants A and B contain the temperature dependent factors. The stress intensity factor can be stated in terms of the applied stress ( $\sigma_a$ ), the crack size (c) and the dimensionless constant (Y) as

$$K = Y \sigma_a \sqrt{c} . (3)$$

<sup>\*</sup> The general concept would also apply for more complex laws, but the detailed results would be different.

Substituting Eq. (3) into Eq. (2),

$$v = \frac{dc}{dt} = B Y^{m} \sigma^{m} c^{m/2} . \tag{4}$$

The total period of creep is limited to the time taken by the initial crack of size  $c_i$  to grow and cause failure. This failure time (tf) is found by integrating Eq. (4) and assuming that  $c_i^{2-m/2} >> c_f^{2-m/2}$  ( $c_f = \text{crack size at failure}$ ):

$$t_{f} = \frac{2 c_{i}^{2-m/2}}{(m-2) B Y^{m}} \sigma_{a}^{-m}$$
 (5)

By substituting Eq. (5) into Eq. (1), the non-elastic strain at failure\* ( $\epsilon_f$ ) is

$$\varepsilon_{f} = \varepsilon_{o} + \frac{2 c_{i} (2-m)/2}{(m-2) Y^{m}} \left(\frac{A}{B}\right) \sigma_{a}^{n-m}$$
 (6)

This equation relates the stress dependencies of creep and slow crack growth to the creep strain at failure.

Likewise the failure time can be related to the steady state creep rate  $(\dot{\epsilon}$  = A  $\sigma_a{}^n)$  by rearranging Eq. (5) and substituting for the stress

$$t_f = A^{m/n} \left( \frac{2 c_i (2-m)/2}{(m-2) B Y^m} \right) \dot{\epsilon}^{-m/2}$$
 (7)

In this equation, creep and slow crack growth are interrelated in terms of creep rate and failure time.

<sup>\*</sup> Since Eq. (1) is derived from steady state creep data, Eq. (6) neglects the strain during tertiary creep.

# 3. DISCUSSION

# 3.1 The Extrinsic Parameters, ci and Y

Equations (6) and (7) show the interrelation between creep and slow crack growth. Both equations were derived by substituting the applied stress-time to failure relation obtained from a fracture mechanics analysis into the commonly used relations to define creep strain. Both equations contain the intrinsic material parameters that define creep strain, viz.,  $\epsilon_0$ , A and n, and those that define the fracture mechanics of slow crack growth, viz., B and m. The two extrinsic parameters are the size of the initial pre-existing crack,  $\epsilon_1$  and the dimensionless constant, Y.

A problem arises concerning the size of the pre-existing crack when the analysis as summarized in Eqs. (6) and (7) is applied to either ductile (metals) or brittle (e.g., ceramics) materials. With regard to metals, the analysis assumes that a crack pre-exists prior to stressing. It is a common opinion among experimentalists investigating the creep of metals that cracks do not pre-exist in metal specimens, but they form after a period of extensive creep strain. The analysis is, therefore, not applicable for this case. On the other hand, the analysis does apply to large metal structures containing large pre-existing cracks, large flaws in castings, welded plates, etc.

Cracks pre-exist in most ceramics. Their distribution of sizes are responsible for the wider distribution of strength values for ceramics relative to metals. [6] Unfortunately, because the sizes of pre-existing cracks in most ceramics are small, techniques do not presently exist to determine their location and to measure their size. The problem of measuring the initial crack size for ceramics can be circumvented by using either the distribution of strength values to define the distribution of crack sizes or by truncating the distribution of crack sizes with a proof test. These two approaches are used in the Appendix to rederive both Eqs. (6) and (7).

A third approach to circumvent the crack size measurement is to simply use the mean strength value to define the mean crack size:

$$\overline{c}_{i} = \frac{\kappa_{c}^{2}}{\gamma^{2} \overline{\sigma^{2}}}, \qquad (8)$$

where  $\overline{\sigma}$  is the mean strength of the material obtained under conditions were slow crack growth does not occur.\*  $K_C$  is the critical stress intensity factor.

Substituting Eq. (8) into Eqs. (6) and (7), one obtains the expressions relating the mean non-elastic strain at failure to the applied stress:

$$\overline{\varepsilon}_{\mathbf{f}} = \varepsilon_{0} + \frac{2}{(m-2) Y^{2}} (\overline{\zeta}_{0})^{(m-2)} (\overline{\zeta}_{0})^{(m-2)} (\overline{\zeta}_{0})^{n-m}$$
(9)

and relating the mean failure time to the creep rate:

$$\overline{t}_{f} = A^{m/n} \left( \frac{2}{(m-2) Y^{2} B} \right) \left( \frac{\overline{\sigma}}{K_{C}} \right)^{(m-2)} \dot{\varepsilon}^{-m/2}$$
(10)

The value of the dimensionless parameter Y depends on the geometry of the crack (location, shape and size), the size of the crack relative to the size of the body and the stress distribution. For many conditions of tensile loading,  $\sqrt{\pi} \le Y \le 2 \sqrt{\pi}$ . [7]

#### 3.2 Temperature Dependence

The parameters A and B express the temperature dependence of creep and slow crack growth, respectively. Assuming activated processes,  $A = A_0 \ e^{-Qc/RT}$  and  $B = B_0 \ e^{-Qcg/RT}$ , where  $Q_C =$  the activation

<sup>\*</sup> For example, if slow crack growth is caused by a corrosive ambient,  $\sigma$  is obtained by measuring strengths in a non-corrosive ambient. Also,  $\sigma$  can be obtained by measuring strengths using very high loading rates to minimize the effect of slow crack growth on strength and thus, the calculation of the initial crack size.

energy for creep and  $Q_{\text{cg}}$  = the activation energy for crack growth.  $A_{\text{O}}$  and  $B_{\text{O}}$  are temperature and stress independent.

If the same phenomenon governs both creep and slow crack growth, the temperature dependence for both will be the same, e.g.,  $Q_c = Q_{cg}$ . For this case, it can be seen by examining Eqs. (9) and (10) that the relation between  $\varepsilon_f$  and  $\sigma_a$  will be independent of temperature, whereas  $t_f$  vs  $\dot{\epsilon}$  will be temperature dependent.

## 3.3 Use of Interrelations

Linear representations for Eq. (9) and (10) can be obtained by taking logarithms. The graphical representations of log  $\epsilon_{\bf f}$  vs log  $\sigma_{\bf a}$  will have a slope of (n-m) and log tf vs log  $\dot{\epsilon}$  will have a slope of -m/n, (see Fig. 1).

From a scientific viewpoint, data plotted in this manner can be instructive for determining whether or not slow crack growth is responsible for failure and for determining the stress exponents for either creep or slow crack growth. For example, m > n for most materials, thus a negative slope for the log  $\epsilon_f$  vs log  $\sigma_a$  strongly suggests that slow crack growth is responsible for failure. If one of the stress exponents is known, the other can be calculated from the relation: slope = n-m. Also, it can be strongly inferred that the same mechanism is responsible for both creep and slow crack growth if the log  $\epsilon_f$  vs log  $\sigma_a$  is independent of temperature.

From an engineering viewpoint, the above analysis shows that strain measurements can be used to predict failure times for materials that concurrently creep and exhibit slow crack growth.\* This is easily illustrated for a component subjected to a constant load at a fixed temperature. (1) The elastic portion of the measured strain determines

<sup>\*</sup> It should be noted that this analysis also is true for two different materials, one exhibiting creep and the other exhibiting slow crack growth, joined in a series arrangement and subjected to the same load. Strain measurements on the material that creeps can be used to predict failure of the material exhibiting slow crack growth.

the applied stress which, in turn, is used to determine the non-elastic strain at failure from either Eq. (9) or a graphical log  $\epsilon_f$  vs log  $\sigma_a$  plot. When the non-elastic strain approaches  $\epsilon_f$ , the component is removed from service. (2) the measured creep rate ( $\dot{\epsilon}$ ) is used to calculate the failure period ( $t_f$ ) using either Eq. (10) or a graphical log  $t_f$  vs log  $\dot{\epsilon}$  plot. The component is removed from service prior to the calculated failure period.

Likewise, but not as simple, failure strains and failure periods can be predicted from strain measurements for components subjected to conditions where both the load and the temperature are a function of time.

## REFERENCES

- 1. J. D. Landes and J. A. Begley, ASTM STP 590, in press (1974).
- 2. J. D. Landes and R. P. Wei, ASME Pub. 73-MAT-B (1973).
- 3. R. Morrell and K. H. G. Ashbee, <u>J. Mat. Sci. 8</u>, 1253 (1973).
- 4. F. F. Lange, <u>J. Am. Ceram. Soc. 57</u>, 84 (1974).
- 5. F. F. Lange, <u>Deformation of Ceramic Materials</u>, Ed. R. C. Bradt and R. E. Tressler, p. 361, Plenum (1975).
- 6. W. Weibull, J. Appl. Mech. 18, 293 (1951).
- 7. A. S. Tetelman and A. G. Evans, <u>Fracture Mechanics of Ceramics</u>, Ed. R. C. Bradt, D. P. H. Hasselman and F. F. Lange, p. 898, Plenum (1974).
- 8. G. J. DeSalvo, Theory and Structural Design Applications of Weibull Statistics, Rep. WANL-TME 2688, Westinghouse Electric Corp.,
  Astronuclear Laboratory (1970).
- 9. A. G. Evans and S. M. Wiederhorn, Intl. J. Fracture (in press).

#### APPENDIX

# 1. Creep-Crack Growth Interrelation Combined with Strength Statistics

The strength  $(\sigma_c)$  probability of failure (P) relation of a material containing a distribution of crack sizes can be expressed as [6]

$$P = 1 - \exp \left[ - \int \left( \frac{\sigma}{\sigma o} \right)^{W} dV \right]$$
 (1A)

where w and  $\sigma_0$  (Weibull's modulus and a normalizing stress, respectively) are material constants; V is the volume of the specimen or component. For certain stressing modes - component configurations. [8]

$$\int_{V} \left(\frac{\sigma}{\sigma_{0}}\right)^{W} dV = LV \left(\frac{\sigma}{\sigma_{0}}\right)^{W}, \qquad (2A)$$

where L is a loading factor, e.g., for pure tension, L = 1.0. Thus, Eq. (1A) can be rewritten as

$$P = 1 - \exp \left[-LV \left(\frac{\sigma}{\sigma_0}\right)^W\right]. \tag{3A}$$

Substituting into this equation the expression relating strength to crack size,

$$\sigma_{c} = \frac{K_{c}}{Y c_{i}^{1/2}},$$

one can obtain<sup>[9]</sup>

$$c_i = \left[\frac{\ln \left(\frac{1}{1-P}\right)^{-2/w}}{LV}\right] \left(\frac{K_c}{\sigma_o Y}\right)^2$$
 (4A)

Substituting Eq. (4A) into Eqs. (6) and (7) of the text, one obtains

$$\varepsilon_{f} = \varepsilon_{o} + \frac{2}{(m-2) Y^{2}} \left[ \frac{\ln \left(\frac{1}{1-P}\right)}{LV} \right]^{\frac{m-2}{W}} \left( \frac{\sigma_{o}}{K_{c}} \right)^{m-2} \left( \frac{A}{B} \right) \sigma_{a}^{n-m}$$
 (5A)

and

$$t_{f} = \frac{2 A^{m/n}}{(m-2) Y^{2} B} \left[\frac{\ln \left(\frac{1}{1-P}\right)}{LV}\right]^{\frac{m-2}{W}} \left(\frac{\sigma_{o}}{K_{c}}\right)^{m-2} \dot{\epsilon}^{-m/n}$$
 (6A)

## 2. Creep-Crack Growth Interrelations Combined with Proof Testing

The maximum crack size in a component that survives an overload proof stress\* (i.e.,  $\sigma_p > \sigma_a$ , where R =  $\sigma_p/\sigma_a$ ) is [9]

$$(c_i)_{\max} = (\frac{K_c}{Y \sigma_p})^2 \le (\frac{K_c}{Y R \sigma_a})^2.$$
 (7A)

Substituting Eq. (7A) into Eqs. (6) and (7) of the text, one obtains a relation for the minimum\*\* creep strain at failure:

$$\varepsilon_{f} \ge (\varepsilon_{f})_{\min} = \varepsilon_{o} + \frac{2}{(m-2) Y^{2}} \left(\frac{R}{K_{c}}\right)^{m-2} \left(\frac{A}{B}\right) \sigma_{a}^{n-2}$$
 (8A)

and the minimum\*\* failure time:

$$t_{f} \ge (t_{f})_{min} = \frac{2 A^{m/n}}{(m-2) Y^{2} B} (\frac{R \sigma_{a}}{K_{C}})^{m-2} \dot{\epsilon}^{-m/n}$$
 (9A)

#### 3. Graphical Representation of Resulting Expressions

As discussed in the text, Eqs. (5A), (6A), (8A) and (9A) can be graphically represented as log-log plots to facilitate the assessment of reliability. The log-log plots formed by these equations are a family of parallel lines shown by the following example for Eqs. (6A and 9A):

$$\log t_{f} = \log \left[ \frac{2 A^{m/n}}{(m-2) Y^{2} B} (\frac{\sigma_{o}}{K_{c}})^{m-2} \right] + \frac{m-2}{w} \log \left[ \frac{\ln \left(\frac{1}{1-P}\right)}{LV} \right] - \frac{m}{n} \log \varepsilon$$

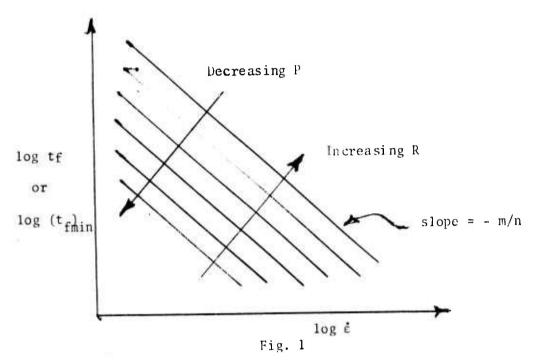
and

$$\log (t_f)_{\min} = \log \left[ \frac{2 A^{m/n}}{(m-2) Y^2 B} (\frac{a}{K_c})^{m-2} \right] + (m-2) \log R - \frac{m}{n} \log \hat{\epsilon}$$

<sup>\*</sup> The same stress distribution must exist in proof testing as in service.

<sup>\*\*</sup> Proof testing only defines the largest possible crack and thus the minimum failure period (and thus, the minimum creep strain at failure).

Plotting these equations as log  $t_f$  ( or log  $(t_f)_{\text{min}}$ ) vs log  $\dot{\epsilon}$ :



These plots are used as follows. For a strain rate measured after a given period of service, the failure time is located on the graph either for a chosen probability of failure, P or, if the component had been proof tested prior to service, for the proof test ratio, R. The remaining service time is then calculated by subtracting the current service time from the failure time.

			# 4		
Unclassified Security Classification					
DECLUSE OF THE COLUMN	ROL DATA - R&D	)	a averall report is classified)		
(Socurity classification of title, body of abstract and indexing annotation must be ent  1 ORIGINATING ACTIVITY (Corporate author)  Westinghouse Research & Development Center  Pittsburgh, PA 15235		2ª REPORT SECURITY CLASSIFICATION Unclassified 2b GROUP			
4. DESCRIPTIVE NOTES (Type of report and inclusive dates)					
5 AUTHOR(S) (Last name, first name, initial)					
Lange, F. F.					
6. REPORT DATE	78. TOTAL NO. OF F		76. NO. OF REFS		
N00014-74-C-0284	98. ORIGINATOR'S REPORT NUMBER(S)				
c.	9b. OTHER REPORT this report) None	NO(S) (Any	other numbers that may be sesigned		
Reproduction in whole or in part is p U. S. Government. Distribution of th	ermitted for ais document i	any pur s UNLIM	pose of the ITED.		
11. SUPPL EMENTARY NOTES	12. SPONSORING MI	LITARY ACT	TIVETY		
	Advanced Research Projects Agency				
13. ABSTRACT	anoon str	ain and	slow crack growth		
Phenomenological interrelations between are derived based on the fact that the determined by the growth kinetics of	he total cree a pre-existi	p strain ng cracl	. It is shown		
that measurements of the creep strai	e periods for	materia	ls that concurrently		
exhibit creep and slow crack growth.					

14 KEY WORDS	LIN	KA	LINK B		LINKC		
	ROLE	wT	ROLE	wT	ROLE	WT	
creep							
slow							
fatigue			:				
failures							
cracks							
growth							

#### INSTRUCTIONS

- 1. ORIGINATING ACTIVITY: Enter the name and address of the contractor, subcontractor, grantee, Department of Defense activity or other organization (corporate author) issuing the report.
- 2a. REPORT SECURITY CLASSIFICATION: Enter the overall security classification of the report. Indicate whether "Restricted Data" is included. Marking is to be in accordance with appropriate security regulations.
- 2b. GROUP: Automatic downgrading is specified in DoD Directive 5200. 10 and Armed Forces Industrial Manual. Enter the group number. Also, when applicable, show that optional markings have been used for Group 3 and Group 4 as authorized.
- 3. REPORT TITLE: Enter the complete report title in all capital letters. Titles in all cases should be unclassified. If a meaningful title cannot be selected without classification, show title classification in all capitals in parenthesis immediately following the title.
- 4. DESCRIPTIVE NOTES: If appropriate, enter the type of report, e.g., interim, progress, summary, annual, or final. Give the inclusive dates when a specific reporting period is covered.
- 5. AUTHOR(S): Enter the name(s) of author(s) as shown on or in the report. Enter last name, first name, middle initial. If military, show rank and branch of service. The name of the principal author is an absolute minimum requirement.
- 6. REPORT DATE: Enter the date of the report as day, month, year; or month, year. If more than one date appears on the report, use date of publication.
- 7a. TOTAL NUMBER OF PAGES: The total page count should follow normal pagination procedures, i.e., enter the number of pages containing information.
- 7b. NUMBER OF REFERENCES: Enter the total number of references cited in the report.
- 8a. CONTRACT OR GRANT NUMBER: If appropriate, enter the applicable number of the contract or grant under which the report was written.
- 8b. 8c, & 8d. PROJECT NUMBER: Enter the appropriate military department identification, such as project number, subproject number, system numbers, task number, etc.
- 9a. ORIGINATOR'S REPORT NUMBER(S): Enter the official report number by which the document will be identified and controlled by the originating activity. This number must be unique to this report.
- 96. OTHER REPORT NUMBER(S): If the report has been assigned any other report numbers (either by the originator or by the sponsor), also enter this number(s).
- AVAILABILITY/LIMITATION NOTICES: Enter any Ilmitations on further dissemination of the report, other than those

imposed by security classification, using standard statements such as:

- "Qualified requesters may obtain copies of this report from DDC."
- (2) "Foreign announcement and dissemination of this report by DDC is not authorized."
- (3) "U. S. Government agencies may obtain copies of this report directly from DDC. Other qualified DDC users shall request through
- (4) "U. S. military agencies may obtain copies of this report directly from DDC. Other qualified users shall request through
- (5) "All distribution of this report is controlled. Qualified DDC users shall request through

If the report has been furnished to the Office of Technical Services, Department of Commerce, for sale to the public, indicate this fact and enter the price, if known.

- 11. SUPPLEMENTARY NOTES: Use for additional explanatory notes.
- 12. SPONSORING MILITARY ACTIVITY: Enter the name of the departmental project office or laboratory sponsoring (paying for) the research and development. Include address.
- 13. ABSTRACT: Enter an abstract giving a brief and factual summary of the document indicative of the report, even though it may also appear elsewhere in the body of the technical report. If additional space is required, a continuation sheet shall be attached.

It is highly desirable that the abstract of classified reports be unclassified. Each paragraph of the abstract shall end with an indication of the military security classification of the information in the paragraph, represented as (TS), (S), (C), or (U).

There is no limitation on the length of the abstract. However, the suggested length is from 150 to 225 words.

14. KEY WORDS: Key words are technically meaningful terma or short phrases that characterize a report and may be used as index entries for cataloging the report. Key words must be selected so that no security classification is required. Identifiers, such as equipment model designation, trade name, military project code name, geographic location, may be used as key words but will be followed by an indication of technical context. The assignment of links, rules, and weights is optional.

Unclassified

CRACK EXTENSION AND ARREST: THEORY AND EXPERIMENTS FOR CONTACT STRESS FIELDS

Technical Report #5, October 15, 1975

Westinghouse Electric Corporation Research and Development Center

Contract Number N00014-74-C-0284

Sponsored by the Advanced Projects Agency ARPA Order Number 2697 Program Code Number 01269

Scientific Officer: Dr. A. M. Diness

Office of Naval Research

Principal Investigator: Dr. F. F. Lange

(412) 256-3684

Effective Date of Contract: April 1, 1974

Contract Expiration Date: June 30, 1976

Amount of Contract: \$159,892

Form Approved, Budget -- No. 22-R0293

The views and conclusions contained in this document are those of the authors and should not be interpreted as necessarily representing the official policies, either expressed or implied, of the Advanced Research Projects Agency of the U. S. Government.

# CRACK EXTENSION AND ARREST: THEORY AND EXPERIMENTS FOR CONTACT STRESS FIELDS

F. F. Lange Materials Science Section

METALLURGY AND METALS PROCESSING DEPARTMENT Westinghouse Research Laboratories Pittsburgh, Pennsylvania 15235

#### ABSTRACT

No.5 Sudden crack extension and arrest is observed when indenters are pressed into the surface of brittle materials. The energitics of this system is examined to determine the criteria for crack extension and arrest. Crack extension is defined by a condition of decreased free energy (after A. A. Griffith) and crack arrest is defined by a condition of increased free energy. The analysis results in two criteria for crack extension. First, a criterion concerning the size of the stress field must be satisfied. This size effect criterion explains the empirical Auerbach's Law which shows that the apparent strength of a brittle material increases with the decreasing size of the contact stress field. Second, a criterion concerning the crack size-maximum tensile stress must be satisfied. Once these two criteria are satisfied, a pre-existing crack will extend and then it will arrest. The size of the arrested crack is related to the size of the pre-existing crack. Experimental observations are presented to confirm the first criterion of crack extension and its relation to material properties. The experiments were performed on hot-pressed  $Si_3^{\text{N}} N_4^{\text{N}}$  and SiC.

#### 1. INTRODUCTION

From a thermodynamic viewpoint, crack extension will only occur when the change in free energy of a cracked body and loading system is equal or less than zero as originally defined by Griffith. To illustrate this criterion, consider the energy (U) expression for the penny-shaped crack in an infinite body as determined by Sack: (1)

$$U = 2\pi c^2 \gamma - \frac{8\sigma^2 c^3 (1-\mu^2)}{3E} ,$$

where  $\gamma$  = surface (or fracture) energy, c = crack length,  $\sigma$  = applied stress and, E,  $\mu$  are the elastic constants of the body. As schematically illustrated in Fig. 1, Griffiths criterion for crack extension is satisfied when the maximum in the U vs c function coincides with the size of the pre-existing crack. As illustrated, this maximum (defined by  $\frac{dU}{dc}$  = 0 and  $\frac{d^2U}{dc^2} < 0$ ) shifts to lower values of c as the stress is increased. When the maximum is shifted to coincide with the size of the pre-existing crack (in this case  $\mathbf{C}_c$ ), the free energy of the system can decrease by crack extension. Once this condition is satisfied for the above energy expression, crack extension is catastrophic because  $\frac{d^2U}{dc^2}$  is always negative for larger crack sizes. The above expression negates the possibility of crack arrest because it does not include a minimum in the U vs c function (defined by  $\frac{dU}{dc}$  = 0 and  $\frac{d^2U}{dc^2} > 0$ ).

The criterion for crack arrest, which is one topic of this paper, must be similar to the fundamentals for crack extension. Just as crack extension is defined by a condition of decreased free energy, crack arrest must be defined by a condition of increased free energy for further crack extension. That is, if a crack is observed to suddenly extend and then to arrest, the U vs c function must contain both a maximum and a minimum. The maximum would occur at lower crack sizes and the minimum at larger crack sizes. Their separation defines the amount of allowable crack extension.

Sudden crack extension and arrest is a common observation, e.g., the thermal cracking of brittle materials. The subject of this paper, viz., crack extension and arrest in contact stress fields, is less common but it is important to the fields of erosion and ballistic impact, surface damage inflicted to brittle materials and brittle materials bearing design.

# 2. CONTACT STRESS FIELDS; INTERACTION WITH BRITTLE MATERIALS

Contact stresses arise whenever two surfaces of finite radii are pressed together. A typical example is the contact stress field that

arises whenever a ball bearing is loaded onto a bearing surface. The distribution of these highly localized stresses, termed Hertzian stresses, both within the bearing and within the bearing surface volume have been well described in the literature (see, e.g., Ref. 2). Their magnitude depends on the applied load (P), the elastic properties of the material (Young's modulus, E and Poisson's Ratio,  $\mu$ ), and the size of the contact area (defined by the diameter D) formed between the bearing and the bearing surface. For the case where both the bearing and bearing surface materials are elastic, the size of the contact area will depend on the applied load and the elastic properties of the materials. For the case where either of the two materials exhibit plastic deformation, the size of the contact area will depend on the applied load and the yield stress of the deformable material.

The stresses of greatest concern here are those that arise in the bearing surface material caused by a spherical indenter. A triaxial state of compression exists in a tear-drop volume beneath the contact area as shown in Fig. 1a. Tensile stresses exist outside of this tear-drop shaped volume. The maximum tensile stress  $(\sigma_{\rm m})$  arises on and is normal to the periphery of the circular contact area:

$$\sigma_{\rm m} = \frac{2(1-2v_{\rm b})P}{\pi D^2}$$
 (1)

where  $v_b$  = Poisson's ratio of the bearing material.

The interaction of contact stresses with metals and ceramics are quite different. Metals will deform in either compression or tension once their elastic limit is exceeded. Ceramics on the other hand, will not deform, but once the load reaches a critical value a large surface crack will suddenly appear at the periphery of the contact area, i.e., within the region of localized tensile stresses. It extends into the material and then arrests. Further crack extension requires an increased load. For the case of spherical indenters, the crack extends beneath the surface in the shape of a cone.

The most interesting fact concerning this cracking phenomenon, as first reported by Auerbach in 1891, (3,4) is that the calculated maximum tensile stress at which spontaneous crack extension occurs is not a constant of the material, but increases with decreasing contact area. That is, a brittle material appears stronger when smaller contact areas are used to introduce a cone crack. This phenomenon is best illustrated by the interaction of contact stresses with glass, a material with a yield stress much greater than its usually measured fracture stress. When a diamond hardness indenter is used to develop contact stresses, glasses will exhibit gross deformation prior to any cracking. (5)

The object of this article is to present a theoretical treatment and experimental data which might help to explain Auerbach Law in terms of the thermodynamics of crack extension and arrest and at the same time, direct engineers in choosing proper brittle materials to be used for applications where a high resistance to surface damage induced by contact stresses is desirable.

# 3. THEORY OF CRACK EXTENSION AND ARREST IN HIGH LOCALIZED STRESS FIELDS

It will be assumed that the bearing surface contains many small, pre-existing cracks and that the size of these cracks (c) are much smaller than the diameter (D) of the contact area formed with a spherical indenter, i.e.,  $c \ll D$ .

If one of the pre-existing cracks, which is located at the periphery of the contact area and favorably oriented with respect to the tensile stresses, is allowed to extend into the Hertzian stress field, the initial stored strain energy  $U_{\overline{SE}}$  in both the bearing surface and the spherical indenter will be reduced due to the diminished stresses during crack extension. The strain energy associated with the contact stress field for a given normalized crack length  $\mu$  = c/D can be expressed as

$$U_{SE} = U_{SE}^{\circ} f(\mu), \qquad (2)$$

where  $f(\mu)$  is a dimensionless function and by definition,  $1 \geq f(\mu) \geq 0$ . Roesler has shown that the initial stored strain energy  $(U_{SE}^{\circ})$  in the system can be expressed as:

$$U_{SE}^{\circ} = \frac{\pi^2 k}{4(1-2v_b)} \sigma_m^2 D^3$$
 (3)

where  $k = \frac{1-\nu_i^2}{E_i} + \frac{1-\nu_b^2}{E_b}$  and  $E_i$ ,  $E_b$ ,  $\mu_c$ ,  $\mu_b$  are the elastic moduli and Poisson's ratios of the indenter and bearing materials, respectively. Substituting Eq. (3) into Eq. (2), one obtains

$$U_{SE} = \frac{\pi^2}{4(1-2\nu_h)} k \sigma_m^2 D^3 f(\mu)$$
 (4)

for the stored strain energy in the system as a function of the normalized crack depth ( $\mu$ ), the maximum tensile stress ( $\sigma_m$ ) and the contact area diameter (D).

The energy required to create the new crack surfaces is

$$U_{S} = \gamma A, \qquad (5)$$

where  $\gamma$  = the fracture energy of the bearing surface material and A = the surface of the crack. U<sub>S</sub> can be expressed as a multiple of the contact area diameter  $(\pi D^2/4)$  as:

$$U_{S} = \frac{\pi}{4} \gamma D^{2} g(\mu), \qquad (6)$$

where  $g(\mu)$  is a dimensionless function of the normalized crack length. Again, by definition,  $g(\mu) \ge 0$ .

Neglecting other energy terms which are assumed not to change during crack extension, the total energy associated with the system for a given relative crack depth is

$$U_{T} = U_{SE} + U_{S} = \frac{\pi^{2}}{4(1-2\nu_{b})} k \sigma_{m}^{2} D^{3} f(\mu) + \frac{\pi}{4} \gamma D^{2} g(\mu)$$
 (7)

Using the thermodynamic criterion introduced by Griffith, crack extension will only occur when it is accompanied by a free energy change  $\leq 0$ , i.e., for an unstable extension of the normalized crack considered here,

$$\frac{dU_{T}}{d\mu} = \frac{\pi^{2}}{4(1-2v_{b})} k \sigma_{m}^{2} D^{3} f'(\mu) + \frac{\pi}{4} \gamma D^{2} g'(\mu) \leq 0,$$
 (8)

This condition must correspond to a maximum in the U vs  $\mu$  curve where  $d^2U_T/d\mu^2 < 0$ . Likewise, crack arrest will occur when  $dU_T/d\mu = 0$  and  $d^2U_T/d\mu^2 > 0$ . This corresponds to a minimum in the U vs  $\mu$  curve. A third condition may also exist where the U vs  $\mu$  curve exhibits neither a maximum nor a minimum point, but only a single inflection. This arises where  $dU_T/d\mu = 0$  and  $d^2U_T/d\mu^2 = 0$ .

Up to this point, no assumptions have been made concerning the specific functional forms of either  $f(\mu)$  or  $g(\mu)$ . It is beyond the scope of this article to derive their explicit forms. For example,  $f(\mu)$  would require knowledge of the stress redistribution during crack extension, which might require a numerical (e.g., finite element) stress analysis. Without going into the details, the general forms of  $f(\mu)$  and  $g(\mu)$  can be indicated in order to draw important conclusions from the previous analysis.

The function  $g(\mu)$  is the fractional increase in the surface area during crack extension. Since it is observed that the area of the Hertzian cone crack increases during crack extension, the function  $g(\mu)$  will not possess an inflection point, viz.,  $g'(\mu) \geq 0$  and  $g''(\mu) \geq 0$  (' and " denote first and second derivatives with respect to  $\mu$ ).

The function  $f(\mu)$  is defined as the fractional release of stored strain-energy during crack extension. By definition  $f'(\mu)$  is always < 0, i.e., at no point during crack extension can the stored strain energy be regained. In order for the  $U_T$  vs  $\mu$  curve to satisfy the observed cracking phenomenon for contact stresses, viz., catastrophic extension at  $P_C$  and crack arrest, it can be shown <sup>(7)</sup> that  $f(\mu)$  must possess at least one

- 7 -

inflection point.\* This implied property of  $f(\mu)$  allows the  $U_T$  vs  $\mu$  function to possess both a maximum and minimum position which respectively relates to conditions where the crack size is large enough for extension and then becomes too large and must arrest.

The implication that  $f(\mu)$  possesses a single inflection also means that there exists conditions where the  $U_T$  vs  $\mu$  function only exhibits a single inflection without either a maximum or a minimum. As stated above, this condition occurs when  $dU_T/d\mu=0$  and  $dU_T^2/d\mu^2=0$ . By determining the first and second derivatives of  $U_T$  with respect to  $\mu$ , it can be seen that the  $U_T$  vs  $\mu$  curve will only possess a single inflection when

$$\sigma_{\rm m}^2 D < \frac{\gamma}{\pi k (1 - 2\nu_{\rm b})} H(\mu_{\rm o}) \tag{9}$$

where  $H(\mu_0) = -g'(\mu_0)/f'(\mu_0) = -g''(\mu_0)/f''(\mu_0)$  and  $\mu_0$  defines the inflection point. Equation (9) is the limiting condition where crack extension will not occur regardless of the size of the pre-existing crack. Since the right hand side of Eq. (9) is a constant for a given bearing-indenter material couple, the first condition that must be satisfied for crack extension is that

$$\sigma_{\rm m}^2 D \ge {\rm constant}$$
 (10)

It can be shown that Eq. (10) is equivalent to Auerbach's Law.\*\*

As schematically illustrated in Fig. 3, once the first condition for crack extension given by Eq. (10) is satisfied, the  $\rm U_T$  vs  $\mu$  curve

<sup>\*</sup> This assumption is based on the observed mechanics of sudden crack extension and arrest for Hertzian cracks. If  $f(\mu)$  did not possess an inflection, the UT vs  $\mu$  function would only have a minimum position and it could not be used to explain the experimental observation of sudden crack extension.

<sup>\*\*</sup> It has been shown (e.g., Ref. 3) that Auerbach's Law is only valid in the range of small contact areas (i.e., small D). Above a certain size D, the apparent strength  $(\sigma_m)$  is independent of the contact area size. Equation (10) shows that as D  $\rightarrow$  0,  $\sigma_m \rightarrow \infty$  as observed for Auerbach's Law; but as D  $\rightarrow \infty$ , the apparent strength  $(\sigma_m)$  of the material cannot  $\rightarrow$  0. That is, although the strength of a brittle material depends on the volume placed under tensile stress due to the statistical nature of flaw distributions, its strength must have a finite value  $(\sigma_t)$  as, for example, determined by a tensile test. Thus, Eq. (1) is only valid where  $\sigma_m \geq \sigma_t$ .

will possess a maximum and a minimum point with increasing  $\sigma_m^{\ 2}$  D corresponding to values of  $\mu$  where crack extension and crack arrest will occur. By examining Eq. (7), the maximum and minimum will shift to lower and higher values of  $\mu$ , respectively as the product  $\sigma_m^{\ 2}$  D is increased. Thus, once Eq. (10) is satisfied, the size of the pre-existing crack,  $\mu_c$ , will determine the value of  $\sigma_m^{\ 2}$ D where crack instability will occur and it will also govern the final length of the stable crack.

It has been shown that two conditions are required for crack extension. First, the product  $\sigma_m^{\ 2} D$  must be greater than a given value which depends on the elastic properties of the bearing and indenter materials and the fracture energy  $(\gamma)$  of the bearing material. Second, once this condition is satisfied, crack extension will occur when the maximum in the  $U_T$  vs  $\mu$  is shifted by increasing the value of  $\sigma_m^{\ 2} D$  to the value  $\mu_C$  defined by the size of the pre-existing crack. At this value, the crack will extend until its normalized size reaches the minimum position in the  $U_T$  vs  $\mu$  curve. Between these two values of  $\mu$ , the condition for unstable crack extension  $(dU_T/d\mu \le 0)$  is maintained.

## 4. EXPERIMENTAL OBSERVATIONS

Equation (10) predicts that a brittle material's resistance to crack extension due to contact stresses will be directly proportional to its fracture energy,  $\gamma$ . It also justifies, on thermodynamical reasoning, Auerbach's Law, viz., that brittle materials appear stronger when smaller contact areas are used to transfer loads.

In order to obtain insight into the validity of these predictions, contact experiments were performed on three materials, viz., hot-pressed  $\mathrm{Si}_3\mathrm{N}_4$  (Norton HS-130), hot-pressed SiC (Norton Co.) and glass (soda-lime silica, Fisher microscope slides). The first two materials are promising candidates for high temperature structural applications and bearing applications. Glass, the third material, was used because its transparency

13

was important in verifying the relation between the presence of Hertzian cracks and the acoustic emission response used to determine  $P_c$ . As reported elsewhere, the fracture energies of hot-pressed  $\text{Si}_3\text{N}_4$  and SiC are 45 J/m² and 22 J/m², respectively. (8) Although not directly measured for these experiments, the fracture energy for various glasses are  $\sim 3 \text{ J/m}^2$ . (9)

The contact experiments were conducted as follows. Similar size blocks ( $^{\circ}$  1.5 x 2.5 x 2.5 cms) of both Si $_3$ N $_4$  and SiC were diamond cut from much larger blocks purchased from the Norton Co. Flexural strength measurements previously performed on bar specimens cut from the same larger blocks were similar for both materials: (8) Si $_3$ N $_4$  (weak direction): 540 MN/m $^2$ , and SiC: 570 MN/m $^2$ . Each of the two blocks were surface finished with a 320 grit diamond grinding wheel so that the surface damage (size and distribution of cracks) were presumably the same. Since the glass microscope slides were only used to justify an experimental procedure described below, they were used without any surface grinding.

Steel ball bearings and tungsten carbide spheres were used as indenters. Initial experiments showed that  ${\rm Al}_2{\rm O}_3$  single crystal spheres were unsuitable since they would crack prior to the introduction of cracks into the  ${\rm Si}_3{\rm N}_4$  bearing surface. Both the steel and tungsten carbide (cemented with cobalt) were observed to deform at loads required to introduce cracks into either  ${\rm Si}_3{\rm N}_4$  or SiC. Both indenter materials were elastic in the load range required to introduce cone cracks into the glass. Different size indenters, ranging from 0.159 to 0.635 cm, were used to obtain different contact areas. Tungsten carbide spheres (0.159 cm diameter) were required to obtain the smallest contact areas for the  ${\rm Si}_3{\rm N}_4$  since steel spheres of this size flattened without introducing cracks.

A load was applied with the moving crosshead (0.005 cm/min) of an Instron testing machine to the spherical indenter which was in contact with the bearing surface. Thick brass plates separated the bearing block from the load cell and the spherical indenter from the

moving crosshead. Rubber pads were also used as acoustic dampers since an acoustic emission technique was used to detect the introduction of a crack into the bearing surface.

A peizoelectric detector was mounted on the brass block in contact with the spherical indenter. The sensitivity of the electronics used to amplify the emission was set to eliminate background noise. Proof that an emission signal corresponded to the introduction of a Hertzian cone crack was obtained by observing numerous surfaces of the glass specimens just prior and after the application of the critical load required to introduce the crack. The Hertzian crack was only observed after an emission signal had been recorded. Thus as the indenter was loaded onto the bearing surface, the critical load required to introduce the cone crack was identified by the acoustic emission signal.

After each experiment, the flat surface on the spherical indenter, caused by plastic deformation, was photographed through a microscope to allow a more accurate measurement of the contact area diameter, D. As previously shown,  $^{(10)}$  this flat area was the same size as the ring crack observed on the bearing surface of  $\mathrm{Si}_3\mathrm{N}_4$  and  $\mathrm{SiC}$ . A new indenter and a different area of the bearing surface was used for each successive experiment.

Equation (1) was used to calculate the maximum tensile stress  $(\sigma_m)$  required to introduce the Hertzian crack in each experiment. As suggested by Eq. (10),  $\sigma_m^2$  was plotted as a function of 1/D for each of the three materials. This is illustrated in Fig. 4.

# 5. DISCUSSION

Figure 4 illustrates that the tensile stress required to introduce surface cracks into each material increases with decreasing contact area diameter. For the smaller contact areas, these stresses are up to

5-6 times the material's strength as measured by flexural testing of bar specimens. This figure also illustrates that a linear line can be drawn through the data for each material as predicted by Eq. (10). Best agreement for this type of plot was obtained for SiC due to the greater variation in the contact diameters obtained during testing and the smaller amount of scatter in data.

Also predicted by Eq. (10) and observed is that the relative slopes of the three sets of data are in relative agreement with the fracture energies of the three respective bearing surface materials. Best agreement was obtained for the relative slopes of the  $\mathrm{Si}_3\mathrm{N}_4$  and  $\mathrm{SiC}$  data. The ratio of these slopes is 1.9 which corresponds well to the ratio of their fracture energies, 2.0. The agreement between the glass and the two other materials is not as good. This may be due to the much lower elastic modulus of glass which should also be included in Eq. (10), the phenomenon of slow crack growth that occurs in glass which would, in effect, lower its fracture energy and the different surface finish for this material relative to the two other materials.

#### ACKNOWLEDGMENT

This work was supported by the Office of Naval Research under Contract No. N00014-68-C-0323.

# LIST OF FIGURES

- FIG. 1 -- Schematic of the energy (U) vs crack size (c) function for a slit crack in a thin sheet with increasing load.
- FIG. 2 -- (a) Contact area formed by spherical indenter and flat bearing surface. (b) Cross-section of Hertzian cone crack introduced into brittle bearing surface material once the load reaches a critical value, P<sub>C</sub>.
- FIG. 3 -- Schematic of  $U_T$  vs  $\mu$  curves for increasing product  $\sigma_m^2$  D.
- FIG. 4 -- Plot of  $\sigma_m^2$ , calculated from experimental data using Eq. (1), vs 1/D for the three materials examined.

## REFERENCES

- 1. R. A. Sack, Phil. Trans. Roy. Soc. (Lon.) 58, 729 (1946).
- 2. F. C. Frank and B. R. Lawn, Proc. Roy. Soc. 299a, 291 (1967).
- 3. F. Auerbach, Ann. Phys. Chem. 43, 61 (1891).
- 4. J. P. A. Tillett, Proc. Phys. Soc. 69B, 47 (1956).
- 5. E. W. Taylor, <u>Nature 163</u>, 323 (1949); <u>J. Soc. Glass Technol. 34</u>, 69T (1950).
- 6. F. C. Roesler, Proc. Phys. Soc. 69B, 55 (1956).
- 7. F. F. Lange, <u>Fracture Mech. of Ceramics</u>, Vol. 2, ed. by Bradt, Hasselman and Lange, p. 599, Plenum (1974).
- 8. F. F. Lange, Ann. Rev. Mat. Sci., Vol. 4, ed. by Huggins, Bube and Roberts, p. 365, Annual Reviews Inc. (1974).
- 9. S. M. Wiederhorn, <u>J. Amer. Ceram.</u> <u>Soc.</u> <u>52</u>, 99 (1969).
- 10. F. F. Lange, "Task III: Relative Resistance of Dense Si<sub>3</sub>N<sub>4</sub> and SiC to Surface Damage Introduced by Hertzian Contact Stresses", NAVAIR Final Rept., April 15, 1973, Contract No. N00019-72-C-0278.

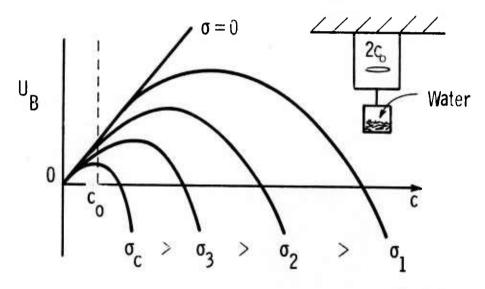


Fig. 1

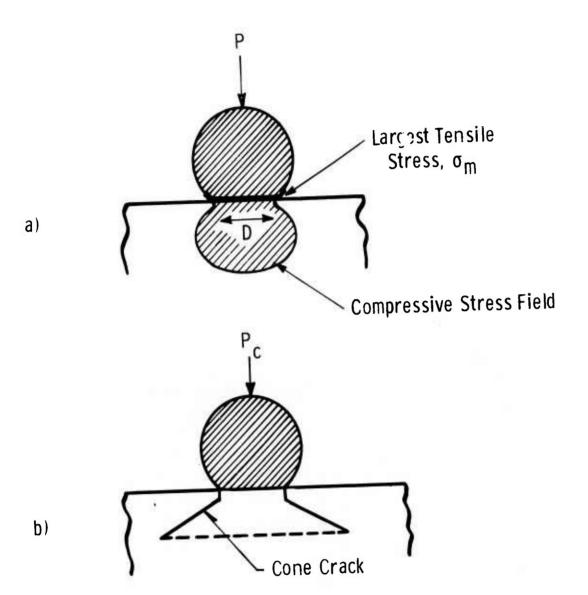


Fig. 2

Curve 677490-A

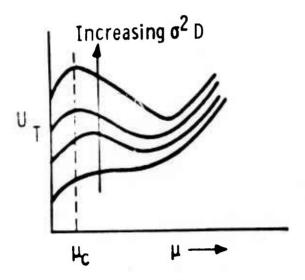
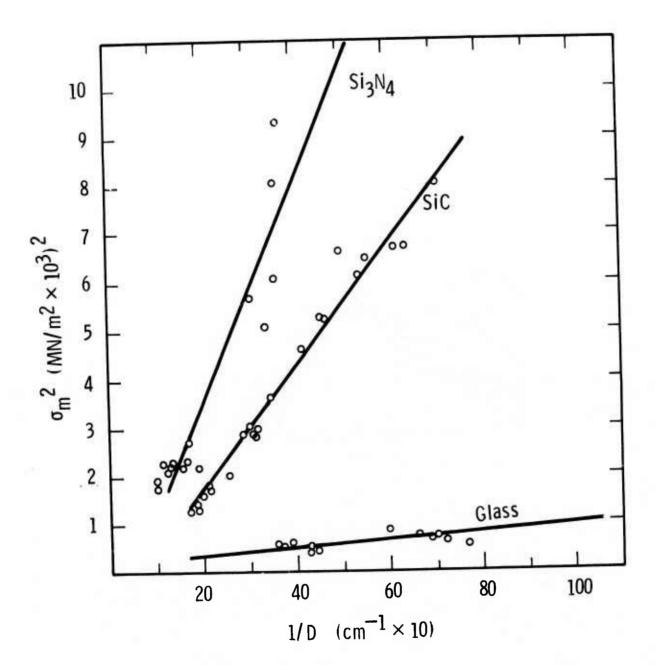


Fig. 3



A ... A

A second College &

A constitution of

Fig. 4

DOCUMENT CON (Security classification of title, body of abstract and indexing	ITROL DATA - R&I	D tered when t	the overall report is classified)			
		28 REPORT SECURITY CLASSIFICATION				
Westinghouse Research & Development Center Pittsburgh, PA 15235		Unclassified				
		26 GROUP	P			
3 REPORT TITLE						
CRACK EXTENSION AND ARREST: THEORY AND	D EXPERIMENTS	FOR CON	TACT STRESS FIELDS			
4 DESCRIPTIVE NOTES (Type of report and inclusive dates)						
5 AUTHOR(S) (Lest name, first name, initial)						
Lange, F. F.						
6 REPORT DATE October 15, 1975	7a. TOTAL NO. OF P	AGES	76. NO. OF REFS			
8 CONTRACT OR GRANT NO.	98. ORIGINATOR'S REPORT NUMBER(S)					
N00014-74-C-0284						
c.	9b. OTHER REPORT this report)	NO(S) (Any	other numbers that may be assigned			
d.	Non e					
10. AVAILABILITY/LIMITATION NOTICES  Reproduction in whole or in part is p	ermitted for a	ייינות אחו	pose of the			
U. S. Government. Distribution of th	is document is	UNLIMI	TED.			
11. SUPPL EMENTARY NOTES	12. SPONSORING MILI	TARY ACT	IVITY			
	Advanced Res	search F	Projects Agency			
13. ABSTRACT						
Sudden crack extension and arrest is						
the surface of brittle materials. Th						
determine the criteria for crack exte	ension and arre	est. C	rack extension is			
defined by a condition of decreased f	free energy (a	fter A.	A. Griffith) and			
crack arrest is defined by a condition	on of increased	d free	energy. The analysis			
results in two criteria for crack ext						
size of the stress field must be sati						
the empirical Auerbach's Law which sl						
material increases with the decreasing						
Second, a criterion concerning the cr						
satisfied. Once these two criteria						
extend and then it will arrest. The						
the size of the pre-existing crack.						
confirm the first criterion of crack						
properties. The experiments were pe						

DD 150RM 1473

Unclassified

extension

Security Classification  KEY WORDS		LINK A		LINK B		LINK C		
		ROLE	wT	ROLE	wT	ROLE	WT	
							= 1	
contacts	3							
fracture	9							
cracks								
bearings	3							
silicon	nitride							
silicon	carbide							
stresses	3	ä						
criterio	on							
Hertz								

#### INSTRUCTIONS

- I. ORIGINATING ACTIVITY: Enter the name and address of the contractor, subcontractor, grantee, Department of Defense activity or other organization (corporate author) issuing the report.
- 2a. REPORT SECURITY CLASSIFICATION: Enter the overall security classification of the report. Indicate whether "Restricted Data" is included. Marking is to be in accordance with appropriate security regulations.
- 2b. GROUP: Automatic downgrading is specified in DoD Directive 5200.10 and Armed Forces Industrial Manual. Enter the group number. Also, when applicable, show that optional markings have been used for Group 3 and Group 4 as authorized.
- 3. REPORT TITLE: Enter the complete report title in all capital letters. Titles in all cases should be unclassified. If a meaningful title cannot be selected without classification, show title classification in all capitals in parenthesis immediately following the title.
- 4. DESCRIPTIVE NOTES: If appropriate, enter the type of report, e.g., interim, progress, summary, annual, or final. Give the inclusive dates when a specific reporting period is covered.
- 5. AUTHOR(S): Enter the name(s) of author(s) as shown on or in the report. Enter last name, first name, middle initial. If military, show rank and branch of service. The name of the principal author is an absolute minimum requirement.
- 6. REPORT DATE: Enter the date of the report as day, month, year, or month, year. If more than one date appears on the report, use date of publication.
- 7a. TOTAL NUMBER OF PAGES: The total page count should follow normal pagination procedures, i.e., enter the number of pages containing information.
- 7b. NUMBER OF REFERENCES: Enter the total number of references cited in the report.
- 8a. CONTRACT OR GRANT NUMBER: If appropriate, enter the applicable number of the contract or grant under which the report was written.
- 8b, 8c, & 8d. PROJECT NUMBER: Enter the appropriate military department identification, such as project number, subproject number, system numbers, task number, etc.
- 9a. ORIGINATOR'S REPORT NUMBER(S): Enter the official report number by which the document will be identified and controlled by the originating activity. This number must be unique to this report.
- 9b. OTHER REPORT NUMBER(S): If the report has been assigned any other report numbers (either by the originator or by the sponsor), also enter this number(s).
- 10. AVAILABILITY/LIMITATION NOTICES: Enter any limitations on further dissemination of the report, other than those

imposed by security classification, using standard statements such as:

- "Qualified requesters may obtain copies of this report from DDC."
- (2) "Foreign announcement and dissemination of this report by DDC is not authorized."
- (3) "U. S. Government agencies may obtain copies of this report directly from DDC. Other qualified DDC users shall request through
- (4) "U. S. military agencies may obtain copies of this report directly from DDC. Other qualified users shall request through
- (5) "All distribution of this report is controlled. Qualified DDC users shall request through

If the report has been furnished to the Office of Technical Services, Department of Commerce, for sale to the public, indicate this fact and enter the price, if known.

- II. SUPPLEMENTARY NOTES: Use for additional explanatory notes.
- 12. SPONSORING MILITARY ACTIVITY: Enter the name of the departmental project office or laboratory sponsoring (paying for) the research and development. Include address.
- 13. ABSTRACT: Enter an abstract giving a brief and factual summary of the document indicative of the report, even though it may also appear elsewhere in the body of the technical report. If additional space is required, a continuation sheet shall be attached.

It is highly desirable that the abstract of classified reports be unclassified. Each paragraph of the abstract shall end with an indication of the military security classification of the information in the paragraph, represented as (TS), (S), (C), or (U).

There is no limitation on the length of the abstract. However, the suggested length is from 150 to 225 words.

14. KEY WORDS: Key words are technically meaningful terms or short phrases that characterize a report and may be used as index entries for cataloging the report. Key words must be selected so that no security classification is required. Identifiers, such as equipment model designation, trade name, military project code name, geographic location, mry be used as key words but will be followed by an indication of technical context. The assignment of links, rules, and weights is optional.

# DISTRIBUTION LIST

Director (2) Advanced Research Projects Agency 1400 Wilson Boulevard Arlington, VA 22209

ATTN: Program Management

Office of Naval Research (5) Department of the Navy Arlington, VA 22217

ATTN: Code 471

Director Office of Naval Research Branch Office 495 Summer Street Boston, MA 02210

Commanding Officer
Office of Naval Research
New York Area Office
207 West 24th Street
New York, NY 10011

Commanding Officer
Office of Naval Research
San Francisco Area Office
50 Fell Street
San Francisco, CA 94102

Director Office of Naval Research Branch Office 1030 East Green Street Pasadena, CA 91101

Commanding Officer Naval Weapons Laboratory Dahlgren, VA 22448

ATTN: Research Division

Commanding Officer Naval Ordnance Laboratory White Oak, Silver Spring, MD 20910

ATTN: Code 210

Director Naval Research Laboratory Washington, D.C. 20390

ATTN: Technical Information
Officer, Code 2000
Code 2020
Code 6000
Code 6100
Code 6300

Code 6400 Code 2029 (ONRL)

Commander
Naval Air Systems Command
Department of the Navy
Washington, D.C. 20360

ATTN: Code AIR 320A Code AIR 5203

Commander
Naval Ordnance Systems Command
Department of the Navy
Washington, D.C. 20360

ATTN: Code ORD 033

Commanding Officer Naval Air Development Center Aeronautical Materials Div. Johnsville Warminster, PA 18974

ATTN: Code MAM

Commander
Naval Ship Systems Command
Department of the Navy
Washington, D.C. 20360

ATTN: Code 0342

Commanding Officer Naval Civil Engineering Laboratory Port Hueneme, CA 93041

ATTN: Code L70

Commander
Naval Ship Engineering Center
Department of the Navy
Washington, D.C. 20360

ATTN: Code 6101

Naval Ships R&D Laboratory Annapolis Division Annapolis, MD 21402

ATTN: Code A800

Commanding Officer
Naval Ships R&D Center
Washington, D.C. 20007

ATTN: Code 747

U. S. Naval Postgraduate School Monterey, CA 93940

ATTN: Dept. of Chemistry and Material Science

Commander Naval Weapons Center China Lake, CA 93555

ATTN: Code 5560

Commander Naval Undersea Warfare Center Pasadena, CA 92152

Scientific Advisor Commandant of the Marine Corps Washington, D.C. 20380

ATTN: Code AX

Commanding Officer
Army Research Office, Durham
Box CM, Duke Station
Durham, NC 27706

ATTN: Metallurgy & Ceramics Div.

Office of Scientific Research Department of the Air Force Washington, D.C. 20333

ATTN: Solid State Div. \$RPS)

National Bureau of Standards Washington, D.C. 20234

ATTN: Metallurgy Division Inorganic Materials Div.

Atomic Energy Commission Washington, D.C. 20545

ATTN: Metals & Materials Branch

Argonne National Laboratory Metallurgy Division P. O. Box 299 Lemont, IL 60439

Dr. T. Vasilos AVCO Corporation Research & Advanced Development Div. 201 Lowell Street Wilmington, MA 01887

Dr. H. A. Perry Naval Ordnance Laboratory Code 230 Silver Spring, MD 20910

Dr. Paul Smith Crystals Branch, Code 6430 Naval Research Laboratory Washington, D.C. 20390

Dr. A. R. C. Westwood RIAS Division Martin-Marietta Corporation 1450 South Rolling Road Baltimore, MD 21227

Dr. R. H. Doremus R.P.I. Materials Engineering Dept. Troy, NY 12181

Professor G. R. Miller Dept. of Ceramic Engineering University of Utah Salt Lake City, UT 84112

Dr. T. D. Chikalla
Fuels & Materials Department
Battelle Northwest
P. O. Box 999
Richland, WA 99352

Mr. G. H. Haertling Ceramic Division Sandia Corp. Albuquerque, NM 87101

Mr. I. Berman
Army Materials & Mechanics
Research Center
Watertown, MA 02171

Dr. R. N. Katz AMMRC Watertown, MA 02171

Professor H. A. McKinstry Pennsylvania State University Materials Research Laboratory University Park, PA 16802

Professor T. A. Litovitz Physics Department Catholic University of America Washington, D.C. 20017

Director (2) Advanced Research Projects Agency 1400 Wilson Boulevard Arlington, VA 22209

ATTN: Program Management

Dr. Harold Liebowitz
Dean of Engineering
George Washington University
Washington, D.C. 20006

Dr. H. Kirchner Ceramic Finishing Company P. O. Box 498 State College, PA 16801

Professor A. H. Heuer Case Western Reserve University University Circle Cleveland, OH 44106

Dr. D. E. Niesz Battelle Memorial Institute 505 King Avenue Columbus, OH 43201 Dr. F. A. Kroger University of Southern California University Park Los Angeles, CA 90007

Dr. S. M. Wiederhorn Inorganic Materials Division National Bureau of Standards Washington, D.C. 20234

Dr. A. G. Evans Rockwell International Science Center P. O. Box 1085 1049 Camino Dos Rios Thousand Oaks, CA 91360

Dr. Co. Hulse United Aircraft Research Labs United Aircraft Corporation East Hartford, CT 06108

Stanford University
Dept. of Materials Sciences
Stanford, CA 94305

Dr. R. K. MacCrone Dept. of Materials Engineering Rensselaer Polytechnic Institute Troy, NY 12181

Dr. D. C. Mattis Belfer Graduate School of Science Yeshiva University New York, NY 10033

Professor R. B. Williamson College of Engineering University of California Berkeley, CA 94720

Professor R. W. Bould
Department of Metallurgical
and Materials Engineering
College of Engineering
University of Florida
Gainesville, FL 32601

Professor V. S. Stubican Department of Materials Science Ceramic Science Section Pennsylvania State University University Park, PA 16802 Professor R. C. Bradt Department of Materials Science Ceramic Science Section Pennsylvania State University University Park, PA 16802

Mr. J. Marshall DCAS District, Pittsburgh 1610-S Federal Building 1000 Liberty Avenue Pittsburgh, PA 15222

Mr. T. H. Lauterschlager Office of Naval Research Arlington, VA 22217

Dr. R. C. Anderson General Electric R&D Center P. O. Box 8 Schenectady, NY 12301

Mr. W. A. Sanders NASA Lewis Research Center Cleveland, OH 44135

Library Stellite Division Cabot Corporation 1020 W. Park Avenue Kokomo, IN 46901

Dr. W. Haller Chief, Inorganic Glass Section National Bureau of Standards Washington, D.C. 20234

Professor M. H. Manghnani University of Hawaii Hawaii Institute of Geophysics 2525 Correa Road Honolulu, HI 96822

Director (2) Advanced Research Projects Agency 1400 Wilson Boulevard Arlington, VA 22209

ATTN: Program Management

Director (6) Naval Research Laboratory Washington, D.C. 20375

ATTN: Code 2629

Director (6)
Naval Research Laboratory
Washington, D.C. 20375

ATTN: Code 2627

Ms. Elizabeth Barrett T/M 3417 TRW Equipment TRW Inc. 23555 Euclid Avenue Cleveland, OH 44117

Brookhaven National Laboratory Technical Information Division Upton, Long Island, NY 11973

ATTN: Research Library

Director Metals and Ceramics Division Oak Ridge National Laboratory P. O. Box X Oak Ridge, TN 37830

Los Alamos Scientific Laboratory P. O. Box 1663 Los Alamos, NM 87544

ATTN: Report Librarian

Commanding Officer
Army Materials and Mechanics
Research Center
Watertown, MA 02172

ATTN: Res. Programs Office (AMXMR-P)

Library
Bldg. 50, Room 134
Lawrence Radiation Laboratory
Berkeley, CA 94720

Commanding Officer Naval Underwater Systems Center Newport, RI 02844 Aerospace Research Laboratories Wright-Patterson AFB Building 450 Dayton, OH 45433

Mr. Richard E. Engdahl President Deposits & Composites, Inc. 1821 Michael Faraday Drive Reston, VA 22090

Defense Metals II. rmation Center Battelle Memorial In. tute 505 King Avenue Columbus, OH 43201

Army Electronics Command Evans Signal Laboratory Solid State Devices Branch c/o Senior Navy Liaison Officer Fort Monmouth, NJ 07703

Commanding General
Department of the Army
Frankford Arsenal
Philadelphia, PA 19137

ATTN: ORDBA-1320, 64-4

Executive Director
Materials Advisory Board
National Academy of Sciences
2101 Constitution Avenue, N.W.
Washington, D.C. 20418

NASA Headquarters Washington, D.C. 20546

ATTN: Code RRM

Air Force Materials Lab Wright-Patterson AFB Dayton, OH 45433

ATTN: MAMC MAAM

Deep Submergence Systems Project Washington, D.C. 20360

ATTN: DSSP-00111

Army Research Office 3045 Columbia Pike Arlington, VA 22204

ATTN: Dr. T. E. Sullivan

Department of Interior Bureau of Mines Washington, D.C. 20240

ATTN: Science & Engineering Advisor

Defense Ceramics Information Center Battelle Memorial Institute 505 King Avenue Columbus, OH 43201

National Aeronautics & Space Adm. Lewis Research Center 21000 Brookpark Road Cleveland, OH 44135

ATTN: Librarian

Naval Missile Center Materials Consultant Code 3312-1 Point Mugu, CA 93041

Commanding Officer Naval Weapons Center Corona Labs. Corona, CA 91720

Commander Naval Air Test Center Weapons Systems Test Div. (Code 01A) Patuxent River, MD 20670

Director Ordnance Research Laboratory P. O. Box 30 State College, PA 16801

Commander
Naval Undersea Warfare Center
271 Catalina Boulevard
San Diego, CA 92152

Director Applied Physics Laboratory Johns Hopkins University 8621 Georgia Avenue Silver Spring, MD 20901

Director Applied Physics Laboratory University of Washington 1013 Northeast Fortieth Street Seattle, WA 98105

Materials Sciences Group, Code S130.1 Navy Electronics Laboratory 271 Catalina Boulevard San Diego, CA 92152

Professor R. Roy Materials Research Laboratory Pennsylvania State University University Park, PA 16802

Professor D. H. Whitmore Department of Metallurgy Northwestern University Evansten, IL 60201

Professor J. A. Pask
Department of Mineral Technology
University of California
Berkeley, CA 94720

Professor D. Turnbull
Div. of Engineering & Applied Science
Harvard University
Pierce Hall
Cambridge, MA 02100

Defense Documentation Center (12) Cameron Station Alexandria, VA 22314

TACTEC
Battelle Memorial Institute
505 King Avenue
Columbus, OH 43201

STOIAC
Battelle Memorial Institute
505 King Avenue
Columbus, OH 43201

Professor D. P. H. Hasselman Materials Research Center Lehigh University, Coxe Lab. Bethlehem, PA 18015

Professor P. E. D. Morgan Franklin Institute Research Labs. 20th & Race Streets Philadelphia, PA 19103

Dr. B. Wilcox Division Director Materials Research National Science Foundation Washington, D.C. 20550

Dr. R. J. Charles General Electric Co. Research & Development Center Box 8 Schenectady, NY 12301

Dr. R. A. Alliegro Norton Company Worcester, MA 01606

Mr. A. F. McLean Ford Motor Company Turbine Research 20000 Rotunda Drive Dearborn, MI 4812

Dr. H. Priest AMMRC Watertown, MA 02171

Mr. J. H. Paster Business Services Mobil Tyco Solar Energy Corp. 16 Hickory Drive Waltham, MA 02154